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Calorimetric measurement of interface enthalpy of nanocrystalline silver(I) oxide (Ag_2O)

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The interface enthalpy of nanocrystalline silver(I) oxide ($\text{Ag}_2\text{O} \cdot n\text{H}_2\text{O}$) was measured. $\text{Ag}_2\text{O} \cdot n\text{H}_2\text{O}$ nanocrystalline samples of varying surface areas and degrees of agglomeration were synthesized by wet chemical technique. Interface areas were estimated by comparing the surface areas measured by N_2 adsorption to the crystallite sizes refined from X-ray diffraction data. The interface enthalpy was verified by utilizing thermodynamic cycle, using enthalpy of solution measurements in 25% HNO_3 at room temperature solution calorimetry. The interface enthalpy of the nanocrystalline $\text{Ag}_2\text{O} \cdot n\text{H}_2\text{O}$ is $(0.842 \pm 0.508 \text{ J/m}^2)$. This work provides the first calorimetric measurement of the interface enthalpy of nanocrystalline silver(I) oxide ($\text{Ag}_2\text{O} \cdot n\text{H}_2\text{O}$).

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