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20th International Conference on

ADVANCED MATERIALS SCIENCE & NANO TECHNOLOGY

August 27-28, 2018 Dubai, UAE

Electrochemical properties of Ni-Zn bi-hydroxides synthesized by hydrothermal methodsteel substrate

N Habib

Université Badji-Mokhtar de Annaba, Algeria

Electrochemical supercapacitors have attracted much interest due to their high power density and longer cycle life. In order to bobtain high performance supercapacitors, many researches are focused on the binary system of transition metal hydroxides and oxides. Therefore, we have synthesized 3D-porous nano/micro-hierarchical Ni-Zn hydroxides using simple and low cost free template hydrothermal process with an important specific surface area, pore volume and size around 100 m2/g, 0.31 cm3/g and 12 nm, respectively. The electrochemical properties of these Ni-Zn based micro-nano-bi-hydroxides have been investigated using Cyclic Voltammetry (CV), galvanostatic Charge/Discharge (CD) and Impedance Spectroscopy (EIS). The electrochemical investigation provides that the binary system transition metal (Ni-Zn) exhibit high electrochemical performance of supercapacitor due to their good electric conductivity, low electronegativity and high electrochemical conductivity.

habib.naima23gmail.com

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