

A Comparison of Silica C and Silica Gel in Hydrophilic Interaction Mode: The Effect of Stationary Phase Surface Area

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Abstract:

Hydrophilic interaction chromatography (HILIC) was first described in 1990 although several interesting reviews focusing on general aspects of HILIC have been published [2-4] and it is becoming increasingly popular, (HILIC) is far from being completely understood.

To assess the effect of silica gel structure on retention in HILIC, a test system was developed using the quaternary ammonium ions benzyltrimethylammonium (BTM), benzyltriethylammonium (BTE) and benzyl dimethylhexylammonium (BDM) as probes with tetramethylammonium acetate (TMAA) as the counter-ion competing against the interaction of the test probes with ionised silanols in the stationary phases. Five columns were examined: 300 Å and 100 Å ACE silica gel columns, 100 Å and 60 Å Kromasil silica gel columns and a Cogent Silica- C column. Retention time was obtained for the test probes at different acetonitrile concentrations with all the mobile phase mixtures containing 10 mM tetramethylammonium acetate (TMAA) buffer at pH [6.00]. In all cases BTE, which has the intermediate ionic radius, was the most retained probe at 20% acetonitrile suggesting that retention on silica gel at low % acetonitrile was due to a balance between lipophilic interaction with siloxane groups and ion exchange. All phases gave “U” shaped plots for log k against percentage of acetonitrile with the steepest rise in retention occurring between 80 and 90% acetonitrile. BTM was the most retained probe at 90% acetonitrile and was most retained on the 60 Å Kromasil column and least retained on the 300 Å ACE silica gel columns.

The ionic strength of the mobile phase was varied at 80 and 90% acetonitrile and plots of log k against the inverse of buffer strength followed by fitting of second order polynomial curves indicated that the largest contribution from HILIC to the mixed HILIC/ ion-exchange mode of retention, in absolute terms, was for the 60 Å silica gel column at 90% acetonitrile and for the Cogent Silica- C column which behaved in most respects like a silica gel column with a pore size around 60 Å. Toluene and pentylbenzene were utilized to survey the diminish in available pore volume due to water assimilation in HILIC mode.

In arrange to evaluate the impact of silica gel structure on maintenance in hydrophilic interaction chromatography; a test

framework was created which utilized quaternary ammonium particles as tests with tetramethylammonium acetic acid derivation (TMAA) as the counter-ion competing against the interaction of the test tests with ionized silanols within the stationary stage. Four silica gel columns and a silica hydride column were inspected. Maintenance times were gotten for the test tests at 20, 40, 60, 80 and 90 % acetonitrile (ACN) with all the versatile stage blends containing 10-mM TMAA buffer at pH 6.0. All stages gave “U”-shaped plots for log k against rate of ACN with the steepest rise in maintenance happening between 80 and 90 % ACN.

Introduction:

The examination of fundamental compounds by fluid chromatography (LC) proceeds to be of intrigued, as over 70% of pharmaceuticals are bases whereas peptides and proteins, which have a few fundamental and corrosive capacities, are a developing course of helpful operators right now being created by numerous pharmaceutical companies. In this way, all the talks are committed to ionizable compounds. Silica is the foremost utilized fabric to get ready stationary stages. Be that as it may, free silanols on the surface of silica are the “villains”, which are mindful for negative intuitive of those compounds and the stationary stage (i.e., terrible top shape, moo effectiveness). For these reasons, a few unused silica and crossover silica stationary stages, which have diminished and/or shielded silanols, are presented within the showcase.

In this setting, Sort C stationary stages are an awfully promising fabric for HILIC, which have an insignificant sum of free silanols, and non-silica stationary stages manage complementary selectivity to silica and cross breed silica stationary phases. Based on the criteria talked about over, this audit is partitioned into three parts: (i) silanol interaction, in which approaches utilized to diminish the sum of free silanols on the surface of silica are looked into; (ii) Sort C silica and HILIC, in which a few applications of this mode and preferences of it over the reversed-phase mode are examined and (iii) non-silica stationary stages, in which the employments of these materials are examined as an elective to silica and half breed silica stationary stages and HILIC applications utilizing non-silica stationary stages are too included.

Silica and Hybrid Silica Stationary Phases:

Silica and Cross breed Silica Stationary Phases When essential compounds are analyzed by RPLC, extra intelligent other than those that are hydrophobic may happen between fundamental compounds and remaining silanols on the surface of the silica . In this case, the compound may be held by ionic intelligent with remaining silanols as well as hydrophobic intelligent with the fortified moiety in a synergistic way, shaping profoundly retentive destinations.

The C18 ligands are as well bulky to respond totally with all silanols; in this way, <50% of the free silanols could be chemically derivatized with C18 and other sorts of natural silanes. Extra receptive silanols can be “endcapped” by response with littler silylating operators such as trimethylchlorosilane, but as numerous as 50% of the initial silanol bunches stay unreacted on a normal

Extended Abstract

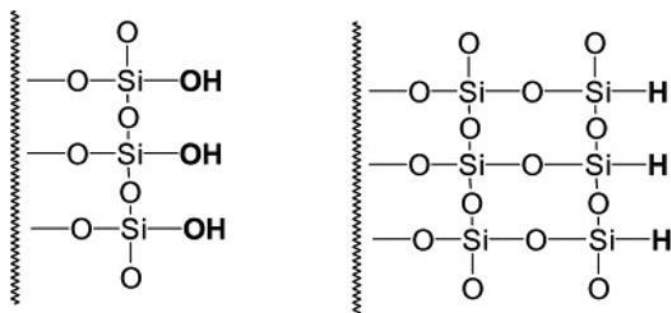
reversed-phase fabric. For this reason, other approaches had to be created to diminish the concentration of free silanols on the stationary stage surface.

The primary approach utilized to decrease the sum of free silanols on the stationary stage surface was the improvement of circular permeable half breed particles for utilize as reversed-phase pressing materials. These stages are made by the co-condensation of methyltriethoxysilane (1 equiv.) with TEOS (2 equiv.) and have an experimental equation of $\text{SiO}_2(\text{CH}_3\text{SiO}_{1.5})_{0.5}$. These methyl cross breed materials shown pore characteristics comparable to those of modern silica-based pressing materials and may well be surface C18 and C8 adjusted utilizing common chlorosilane-bonding conventions. These stationary stages have been commercially accessible since 1999 beneath the exchange title XTerra, and materials arranged utilizing this innovation are alluded to as the first-generation half breed.

Type C Silica:

Silica-hydride materials, which are moreover known as Sort C silica, speak to a unused sort of HPLC stationary stage based on high-purity silica, whereby the surface of this modern fabric is to a great extent populated with the less polar silicon-hydride (Si-H) bunches, rather than the polar silanol bunches (Si-OH) that cover the surface of all other assortments of adjusted silica

The differences between Type B silica and Type C silica are outlined in Figure 2. The records illustrations of silica-hydride stages (e.g., little alkyl, phenyl, cholesterol, octadecyl and alkyl carboxylic acids) that have brought about in noteworthy applications to date. In this fabric, alluded to as silica hydride, Si-H moieties supplant ~95% of the Si-OH bunches on the surface



Sort C silica is one of the foremost promising chromatographic back materials. In any case, this fabric has never come to the utility, breadth and significance of Sort B silica for HPLC divisions. It happens due to its limited number of providers and since Sort C stationary stages are not accessible as core-shell and sub-2 μm stationary stages.

Hydrophilic Interaction Liquid Chromatography:

Hydrophilic ionizable compounds may have moo maintenance on

reversed-phase stationary stages, which require the utilize of versatile stages containing moo natural modifiers (indeed 100% watery portable stages). Thus, stationary stages must be congruous beneath these conditions (i.e., that they don't show stage de-wetting/phase collapse). Stages such as polar implanted and polar endcapped are as often as possible utilized for these tall watery chromatographic conditions. As of late, the strategy of HILIC has gotten to be a progressively prevalent elective strategy for the division of polar compounds when polar stationary stages and versatile stages wealthy in natural solvents (more often than not acetonitrile, ACN) are employed. HILIC could be a variety of typical stage LC that utilizes more eco-friendly and less poisonous solvents than chlorinated solvents and hexane utilized in typical stage fluid chromatography.

The Non-Silica Stationary Phases:

The non-silica stationary stages are another choice for the examination of fundamental pharmaceuticals, since they give selectivity diverse from silica and half breed silica stationary stages.

Conclusion:

The development of modern second-generation half breed silica, Sort C silica, PGC and zirconia polymer-coated stationary stages presently permits the chromatographer to misuse a much more extensive plan space (i.e., temperature and pH), counting soluble versatile stages that were already denied, and consequently the likelihood of fulfilling the required chromatographic selectivity/resolution is expanded. These unused sorts of stationary phases have too been appeared to supply tall efficiencies and fabulous top shape for the chromatography of essential compounds particularly when they are analyzed in their ion-suppressed mode. Lean and symmetrical crests beneath tall and moo ionic-strength portable stages have been illustrated. This property is of major significance in fast generic screening utilizing LC-MS, which frequently utilizes dissolvable exchanging to assess compound immaculateness in moo and tall pH portable stages.