Journal of Nanomaterials & Molecular Nanotechnology

Research

Corrosion Studies of Silicon Carbide (Sic) Reinforced Al 8088 Nano Composites (Ncs) in an Alkaline Medium Using Weight Loss Method

Pruthviraj RD

Department of Chemistry, Raja Rajeswari College of Engineering, Bangalore, India

*Corresponding author: Pruthviraj RD, Department of Chemistry, Raja Rajeswari College of Engineering, Bangalore, India, Tel: +91- 9742832678; E-mail: pruthvirajrd@gmail.com

Abstract

In this investigation, an strive has been made to broaden Al 8088 reinforced with Silicon carbide particulate nano composite through liquid melt metallurgy method and to look at corrosion conduct. nano composite containing 2, 4 and 6 percent of SiC are prepared consistent with ASTM requirements. Weight loss approach is carried out in 0.025N, 0.05N and 0.1N answers of KOH. The corrosion price decreases with boom in the publicity time for all the specimens and decreases with the boom in reinforcement content material of thenano composites. Sincenano composites undergo much less corrosion compared to base alloy, which can be used for the manufacture of equipments used in marine environment.

Keywords: Nano composites, Corrosion, Al 8088, Silicon carbide, Vortex.

Introduction

Technological improvement depends on the advances within the subject of materials. The call for made on substances forever better overall performance are so exceptional and various that no one fabric is capable of fulfill the varying demands from the distinct industries. This naturally brought about the concept of combining specific substances in an quint essential nano composite to satisfy the consumer requirements [1].

In today's industry, nano composite fabric plays an essential function in manufacturing engineering, utilization and education. Nano composite substances are materials made from two or greater essential materials that remain wonderful and separate whilst forming a unmarried component. Generally, one material offers a non-stop matrix at the same time as the opposite paperwork the reinforcement [2]. The two materials have to be chemically inert with recognize to every other so no chemical interaction happens upon heating until one in all the additives melts, an exception to this circumstance is a small diploma of inter diffusion on the reinforcement-matrix interface to decorate bonding. The two materials get combined with each different very intimately and a mechanical bonding takes vicinity between the mixtures may include metals, polymers or ceramics. Prime engineering goal is to acquire desirable homes from the aggregate of materials [3].

More recently researchers have also all started to actively recognize actuation, computation and verbal exchange intonano composites which are known as robotic substances. Typical engineered nano composite substances includes mortars concrete strengthened plastics consisting of steelnano composites, fiber bolstered polymer nano composites and ceramic nano composites. Nanonano composite and nano composite ceramic materials are generally used for buildings, bridges and structures which includes boat hulls, swimming pool panels , race car bodies , bathe stalls , bathtubs, storage tanks, imitation granites and cultured marbles sinks and countertops and the most superior examples perform robotically on space crafts and aircrafts. Nanonano composites are vital class of substances, which incorporate metallic or alloy as matrix and a ceramic particulate or fiber or whiskers as reinforcements. Aluminium primarily based Nanonano composites show off more advantageous corrosion resistance, put on and mechanical houses. They provide significantly enhanced residences over metals and alloys [4].

Materials and Methods

The material selected for the present research paintings is popularly used Al 8088 alloy which is commercially available. Its composition is given in desk 1. The reinforcement used in the present studies paintings is 50-80micro M size Silicon Carbide particulates that is commercially available [5]. The technique used for corrosion characterization is static weight loss corrosion method as in step with ASTM requirements G69-80.

Element	Cu	Cr	Mn	Mg	Si	Ti	Zn	Fe	Al
Percentage	1.8	0.2	0.4	1.9	0.5	0.15	3.25	0.5	Bal

Table 1: Composition of Al 8088 alloy.

Preparation of nano composites by liquid melt metallurgy technique

The nano composites have been prepared *via* liquid soften metallurgy technique *via* the system of Vortex method. Fig 1 is the furnace used for the education of the nano composite is basically an electrically heated 3-section resistance furnace of 12KW potential fitted with three pairs of 14 guage kanthal A1 grade heating coils. The furnace is outfitted with an alumina crucible at its middle with a gap at bottom and melt may be added directly into the mould. The furnace may be tilted at 90 stages on its horizontal axis

Figure 1: Bottom pour furnace.

A muffle furnace was used to preheat the SiC particulates to an superior temperature of about 4000C and maintained at that temperature until it was added into the molten alloy melt. The preheating of the reinforcement is necessary on the way to reduce the temperature gradient among the molten steel and the reinforcement to reap better bonding by way of decreasing difference in the surface energy. AL 8088 alloy has a melting point variety of 5800-6600C. The melt turned into superheated to a temperature of 6300C and maintained at that temperature. A chrome steel impeller / stirrer covered with aluminite coating a good way to save you the migration of ferrous ions from the stirrer material to the melt is introduced in to the soften [7].

The impeller used for stirring was of centrifugal kind with three blades welded at 450 inclination and 1200 apart. The stirrer become circled at a velocity of 500rpm and a vortex changed into created within the soften.

The depth of stirring, and degassing parameters like gasoline flow charge, stress and time of flow had been standardized. This was carried out as basic observe by way of using water module studies, and base steel became used to optimize the opposite parameters for better soundness and mechanical houses. After the addition of the particulates, the soften turned into degassed by means of using natural nitrogen for about 3-4 minutes in to soften and pour immediately in to the preheated mould. Following this procedure, AL 8088/SiC particulatenano composites of 0%, 2%, 4% and 6% by using weight were prepared. The preparednano composites are subjected to evaluation of important weight reduction take a look at [8].

Specimen preparation

Weight loss Test: The specimens were organized from the bar castings. Cylindrical specimen of size 20 mm x 20mm are machined from the bar castings of thenano composites and the matrix alloy. The samples had been successively ground using 220, 320, four hundred and 900 Emery paper and have been very well polished according to metallographic techniques and degreased in acetone and dried.

All the specimens are subjected to conventional metallographic techniques as achieved by S. EzhilVannan and Paul Vizhian Simson earlier than subjecting them to static weight reduction distribution of Silicon Carbide is observed. The samples have been weighed as much as fourth decimal location the use of electronic balance and also the specimen dimensions were cited down the usage of vernirer gauge [9].

Experimental Procedure

Weight loss method (immersion test):

The corrosion behavior of Al 8088 matrix and Al 8088-SiC particulatenano composites have been studied by way of immersion test. The samples have been prepared consistent with ASTM standards. They suspended in 0.025N, 0.05N & zero.1N concentration of KOH for distinctive time intervals as much as ninety six hours in steps of 24 hours. To reduce the contamination of the aqueous solution and loss because of evaporation, the beakers have been blanketed with paraffin paper all through the entire test period. After the required time the samples have been wiped clean mechanically with the aid of the usage of a brush as a way to do away with the heavy corrosion deposits at the floor [10]. The corresponding changes inside the weights had been noted. After the determination of weight reduction C-Programming changed into included for the formula,

Corrosion rate = 534 W/DAT mpy

Where W is the weight reduction in gm, D is density of the specimen in gm/cc, A is the place of the specimen (inch2) and T is the exposure time in hours and corrosion fee is calculated the usage of the equal programme within the computer. In the present studies work corrosion costs in terms of mmpy for the matrix andnano composites have been measured [11].

Results and Discussion

% of weight loss of Al 8088 matrix in different concentration of KOH solution

SiC Content (%)	Time of Exposure			
	24hr	48hr	72hr	96hr
0	0.542	0.439	0.279	0.222
2	0.526	0.402	0.251	0.24
4	0.439	0.364	0.248	0.235
6	0.357	0.324	0.238	0.219

Table 2: % Weight loss test for Al 8088/SiC Specimens in 0.1N KOH.

SiC Content (%)	Time of Exposure			
	24hr	48hr	72hr	96hr
0	0.478	0.427	0.285	0.213
2	0.455	0.405	0.242	0.237
4	0.419	0.348	0.221	0.223
6	0.339	0.313	0.217	0.215

Table 3: % Weight loss test for Al 8088/SiC Specimens in 0.05N KOH.

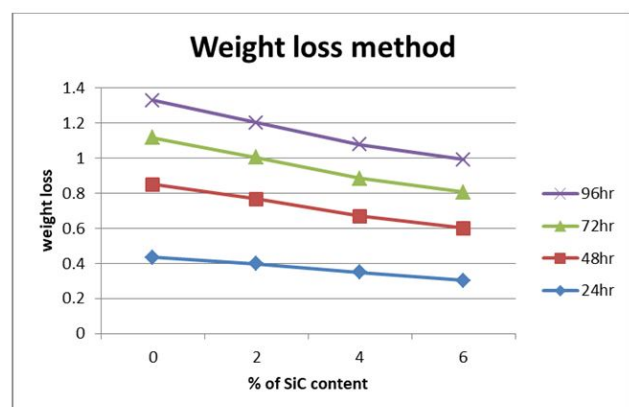| SiC Content (%)	Time of Exposure			
	24hr	48hr	72hr	96hr
0	0.435	0.415	0.267	0.212
2	0.398	0.37	0.235	0.199
4	0.35	0.319	0.214	0.195
6	0.304	0.297	0.204	0.187

Table 4: % Weight loss test for Al 8088/SiC Specimens in 0.025N KOH.
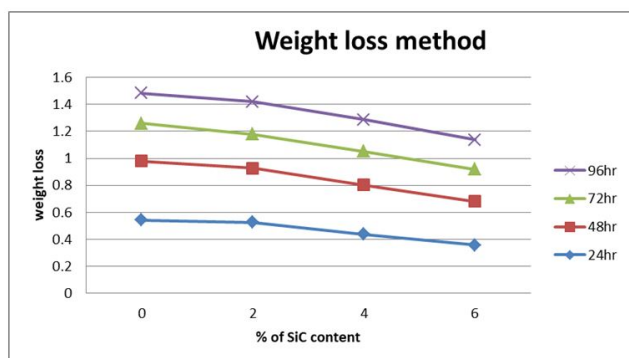
Figure 2: % Weight loss test for Al 8088/SiC Specimens in 0.1N KOH.
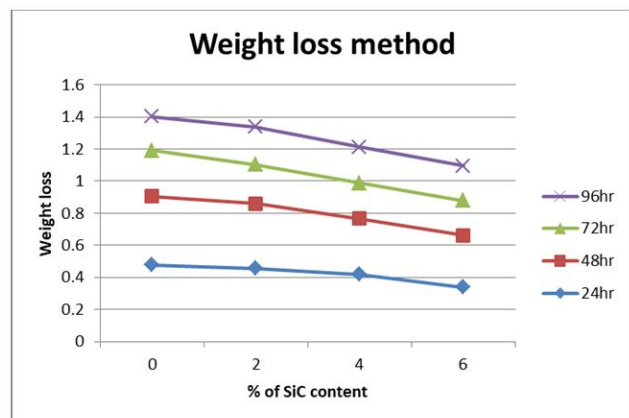
Figure 3: % Weight loss test for Al 8088/SiC Specimens in 0.05N KOH.

Figure 4: % Weight loss test for Al 8088/SiC Specimens in 0.025N KOH.

Effect of corrosion test duration

The consequences in tables 2 to four indicates that, corrosion fee of matrix andnano composite decreases with growth in time of publicity and also resistance ofnano composite to corrosion increases with boom in time of exposure , this is because of the formation of aluminium hydroxide movie on the floor [12]. This layer protects in addition corrosion in alkaline media. But genuine chemical nature of the movie still isn't determined.

Effect of SiC content

The Silicon Carbide content in Al 8088 alloys plays a considerable function within the corrosion resistance of the fabric. Increase in the proportion of Silicon Carbide will be high quality to reduce the density and growth inside the energy of the alloy, however the corrosion resistance is thereby extensively extended. Al 8088 MMCs when strengthened with Silicon Carbide of weight percent from 0 to 6% could be correctly produced *via* liquid soften metallurgy method. The charge of corrosion of both the alloy andnano composite decreased with growth in time duration in all concentrations of KOH solutions as proven in Fig. 2 to four. The corrosion fee of thenano composites become lower than that of the corresponding matrix alloy in concentrations of KOH solution [13].

Conclusion

Based at the result presented, the following conclusions can be drawn
AA 8088-SiCnano composites have been efficiently fabricated the use of liquid melt metallurgy technique. The Al 8088/SiC nano composites exhibited better corrosion resistance than the Al 8088 matrix in all of the concentration of KOH. Increase within the weight percent of SiC particulates, multiplied the corrosion resistance of the Al 8088/SiC nano composites. Increasing the period publicity reduces the corrosion rate of both alloy andnano composite.

Acknowledgments

References

1. Abd Ei-Rehim SS, Ibrahim MAM, Khaled KF (1999) 4-Aminoantipyrine as an inhibitor of mild steel corrosion in HCl solution. J Appl Electrochem 29: 593-599.

2. Almeida E, Pereira D, Figueiredo MO, Lobo VMM, Morcillo M (1997) The influence of the interfacial conditions on rust conversion by phosphoric acid. Corros Sci 39: 1561-1570.

3. Asthana R, Singh M, Sobczak N (2005) The role of wetting and reactivity in infiltration of ceramic-metal composites. Ceram Eng Sci Proc 2: 249-261.

4. Bellanger G, Rameau JJ (1996) Effect of slightly acid pH with or without chloride in radioactive water on the corrosion of maraging steel. J Nuclear Mat 228:24-37.

5. Bellanger G (1994) Effect of carbonate in slightly alkaline medium on the corrosion of maraging steel. J Nuclear Mat 217:187-193.

6. Cheng S, Chen S, Liu T, Chang X, Yin Y (2007) Carboxymethylchitosan + Cu2+ mixture as an inhibitor used for mild steel in 1 M HCL. Electrochim Acta 52: 5932-5938.

7. Castle JE, Sun L, Yan H (1994) The use of scanning auger microscopy to locate cathodic centers in SiC/Al6061 MMC And to determine the current density at which they operate. Corrosion Science 36: 1093-1110.

8. Deve HE, McCullough C (1995) Continuous-fiber reinforced Al composites: A new generation. JOM 4: 33-37.

9. DeSalazar JMG, Urefia A, Mazanedo S, Barrens M (1999) Corrosion behaviour of AA6061 and AA7075 reinforced with Al2O3 particulates in aerated 3.5%chloride solution potentiodynamic measurements and microstructure evaluation. Corrosion Science 41: 529-545.

10. El-Neami KKH, Mohamed AK, Kenawy IM, Fouda AS (1995) Inhibition of the corrosion of iron by oxygen and nitrogen containing compounds. Monatsh Chem J 126: 369-376.

11. Gunasekaran G, Chauhan LR (2004) Eco friendly inhibitor for corrosion inhibition of mild steel in phosphoric acid medium. Electrochim Acta 49: 4387- 4395.

12. Gurrappa VV, Bhanu P (2006) Corrosion characteristics of aluminium based metal matrix composites. Materials Science and Technology. 22: 115-22.

13. Katsuaki Suganuma. (1993) Interfaces in β–SiC whiskers/6061 aluminum composites. Journal of Materials Research 8: 2569-2576.